



Rheology of partially crystallized simulated nuclear glass melts

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Context of the study: motivation for studying the impact of crystals on melt rheology



Context of the study: impact of Pt^{ide} particles on glass melt rheology



French nuclear glass

~ 40 elements in an alumino-borossilicate glass Platinoids => particles in suspension in the glass melt Pd(Rh-Te) beads (1-10 μm) Ru(Rh)O₂ needles (10-50 μm)



(Hanotin et al., 2016)

Typically ~1 vol% of Pt^{ide} particles (∀T) => strong impact on glass melt rheology



Context of the study:

Challenge of the study



Do crystals affect the rheology of the melt as Pt^{ide} particles do?



Challenge of the study



Particularity of crystals compared to Pt^{ides}

Unlike Pt^{ide} particles, crystals are soluble in the melt at $HT \rightarrow their$ fraction depends on t and T. Depending on T, they can grow or dissolve. This (T, t)-dependency can be described by JMAK equations:





1. Why studying the impact of crystals on glass melt rheology ?

2. Experimental method

3. Results





Experimental method: studied system

SYSTEM: melt + apatite

Initial glass composition

%

mass.

36,9

13,8

%

mol.

51,5

16,6

Apatite (Ca₂Nd₈(SiO₄)₆O₂)

Full / hollow hexagonal needles



SEM images of apatites [Renaud, 2017]



T_{liq} = 1178°C

Microtomography of apatites [Delattre, 2014] ESRF, ID19.

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 Na2O
 12,3
 16,6

 CaO
 4,9
 7,4

 Nd2O3
 32,1
 8,0

 Renaud, 2017
 2017



Oxydes

SiO₂

 B_2O_3

Experimental method: studied system



SYSTEM: melt + apatite

Viscosity



[Renaud, 2017]

Initial glass composition

Oxydes	% mass.	% mol.
SiO ₂	36,9	51,5
B_2O_3	13,8	16,6
Na ₂ O	12,3	16,6
CaO	4,9	7,4
Nd_2O_3	32,1	8,0

Renaud, 2017

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Experimental method: investigation of the studied system





1. Why studying the impact of crystals on glass melt rheology ?

2. Experimental method

3. Results: impact of apatites on melt rheology



Results: apatite crystallization in static conditions







Results: impact of dynamic conditions on apatite crystallization







Viscosity once $\Phi_{\rm eq}$ has been reached 10⁵ 35 A SEM Images (%surf, This study) ~2,3 surf% apatite O SEM Images (%surf, Renaud, 2018) Impact of apatites at several 30 EPMA (%vol) Relative Viscosity η_R (η/η_L) $_{0}$ $_{0}$ $_{0}$ $_{0}$ $_{0}$ ~3,4 surf% apatite Data fitting temperatures = several Φ ~4,5 surf% apatite * ~5,7 surf% apatite 25 √ 1100°C ~ 5,7 vol% (%) ⁵⁰ Ø 15 ✓ 1116°C ~ 4,5 vol% **√**1132°C ~ 3,4 vol% ✓ 1147°C ~ 2,3 vol% 10 ✓ 1162°C ~ 1,2 vol% 5 0 L 10⁰ 800 900 1000 1100 1200 Température (°C) 10¹ 10² 10⁰ Stress (Pa) The studied apatite-bearing melt becomes non-newtonian from 3 vol%.

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The theoretical value of Φ_{c} seems to be in agreement with our experimental results.



Cerianite

[Jiusti, 2023a]



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Dependence of the critical particle fraction for a system to present a yield strength as a function of the aspect ratio of the particles (Simulated for shear stress = 0) [Saar, 2007].





In the case of Pt^{ide} particles, particle interaction between the liquid PdRhTe alliage and the $Ru(Rh)O_2$ needles may also play a role

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Conclusion: impact of crystallization on melt rheology

- The rheology of a partially crystallized glass melt depends mostly on:
 - 1) Crystal morphology
 - 2) The crystallized fraction

In both cerianite and apatite cases, the melt can tolerate a higher fraction of crystals before becoming non-Newtonian when compared to Pt^{ide} particles.

- For a few vol% of cerianite => VFT models can be used
- For a few vol% of apatites => the loss of the Newtonian behavior has to be taken into account in models





Conclusion: impact of crystallization on melt rheology



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Modelization during crystal growth / at crystallization equilibrium



- Take into account the relaxation phase (empirical model)
- Developpe a model based on JMAK equation for phases 2 and 3, taking into account the value of the plateau with the applied shear rate (empirical model or model based on $\Phi_{\rm eff}$).



Sisko parameters for the different apatite fractions studied

	Φ = 1.2% (1162 °C)	Φ = 2.3% (1147 °C)	Φ = 3.4% (1132 °C)	Φ = 4.5% (1116 °C)	Φ = 5.7% (1100 °C)
η_{∞} (Pa.s)		1.99 ± 23	3.52 ± 0.06	6.01 ± 0.04	10.20 ± 0.03
k	Newtonian	0.27 ± 37	2.8 ± 0.1	4.71 ± 0.06	9.24 ± 0.07
N		-0.27 ± 112	0.13 ± 0.06	0.15 ± 0.02	0.06 ± 0.01

+ crystal sedimentation has to be studied.

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Thanks for your attention!

